

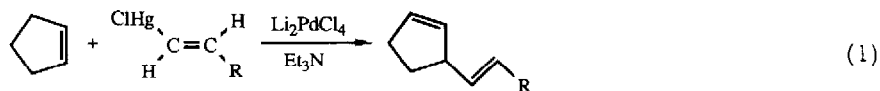
ORGANOPALLADIUM APPROACHES TO PROSTAGLANDINS. 10. AN EFFICIENT SYNTHESIS OF
PROSTAGLANDIN E₂ VIA VINYPALLADATION OF 4-CYCLOPENTENE-1,3-DIOL

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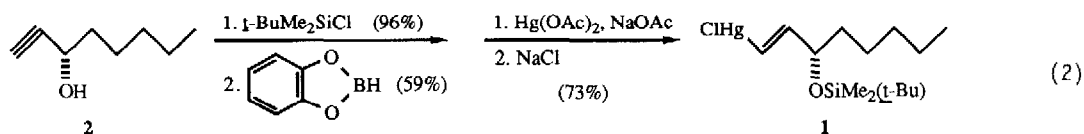
Summary. Prostaglandin E₂ is synthesized by an efficient two step sequence involving (1) preparation of ketol **4** via vinylpalladation of 4-cyclopentene-1,3-diol using either vinylmercurial **1** or vinylic iodide **5**, and (2) subsequent regio- and stereoselective alkylation via sequential dianion generation, tin enolate formation, and organic halide addition.

Prostaglandins are an extremely important, physiologically active class of compounds whose synthesis has received a great deal of attention in recent years.¹⁻³ Most efficient of the present methodology for the synthesis of prostaglandins are three-component coupling processes⁴ involving organocopper conjugate addition to enones⁵⁻¹⁰ and derivatives¹¹ and subsequent trapping of the resulting enolate. We wish to report an entirely new, very efficient, organopalladium-based, three component coupling process for the synthesis of prostaglandins, and more specifically prostaglandin E₂ (PGE₂).

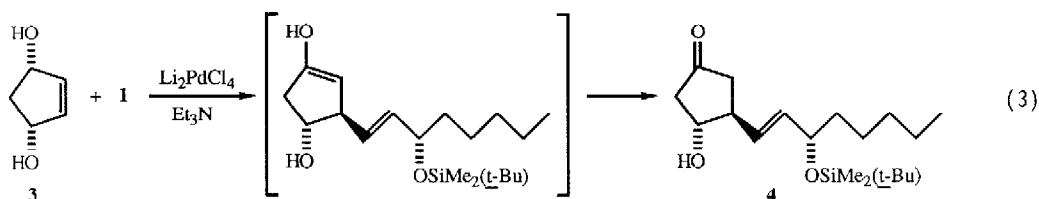
Our approach involves the palladium-promoted vinylation of cycloalkenes.¹² We have recently reported the facile coupling of cycloalkenes and vinylpalladium intermediates derived from vinylmercurials (eq. 1).¹³ We reasoned that analogous chemistry using enantiomerically



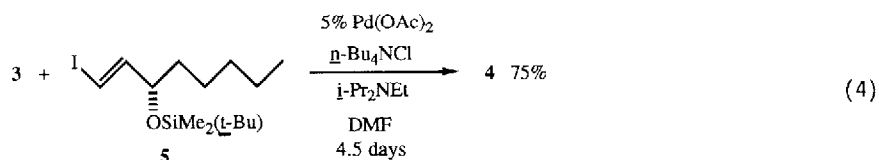
pure vinylmercurial **1**,¹⁴ easily prepared¹⁵ from (*S*)-1-octyn-3-ol (**2**)¹⁶ (eq. 2), and readily available 4-cyclopentene-1,3-diol (**3**)¹⁷ should afford an intermediate appropriately



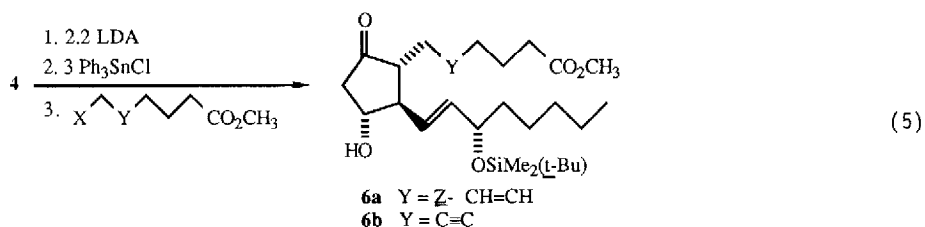
functionalized for further elaboration to PGE₂. Indeed, this reaction¹⁸ affords a 70% isolated, unoptimized yield of hydroxyketone **4**¹⁹ as a 1:1 mixture of diastereomers (from addition to opposite ends of the carbon-carbon double bond) (eq. 3). This reaction appears to be a very general route to ketols such as **4**. However, this approach suffers the obvious disadvantage that one must prepare an organomercurial and the reaction requires stoichiometric amounts of palladium.



Recent work in our laboratories²⁰ and those of others²¹ on the palladium-catalyzed inter- and intramolecular arylation and vinylation of cycloalkenes using the corresponding organic halides suggested that intermolecular vinylation of diol **3** might provide a more convenient route to ketol **4**. In fact, the room temperature reaction²² of enantiomerically pure vinylic iodide **5** [prepared most conveniently from (*S*)-1-octyn-3-ol via silylation and subsequent hydrozirconation-iodination^{23,24}] and diol **3** afforded a diastereomeric mixture of ketol **4** in 75% isolated, unoptimized yield (eq. 4).



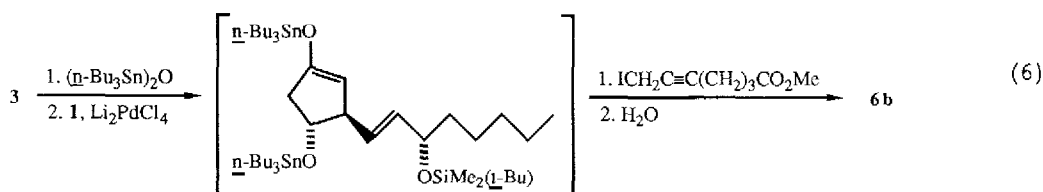
At first glance the regio- and stereospecific alkylation of ketol **4** appears an insurmountable problem. After examining a variety of possible solutions, we have found that treatment of ketol **4** with 2.2 equiv of lithium diisopropylamide (LDA) in THF at -78°C for 60 mins, takes advantage of prior alkoxide formation to regioselectively generate the desired enolate (eq. 5). Direct alkylation of this intermediate did not afford clean products.



However, prior treatment with 3 equiv of Ph_3SnCl ²⁵ (-78°C , 15 min) and low temperature alkylation (5 equiv alkyl halide; HMPA; -78°C for 20 mins, then -30 to -20°C for 7 h) using methyl *cis*-7-bromo-5-heptenoate gave a 17% yield of 11,15-bis(*O*-*tert*-butyldimethylsilyl) PGE₂ methyl ester after further silylation (necessary to separate **6a** from **4**). Using the corresponding, relatively unstable allylic iodide, the yield was raised to 24% (36% yield based on recovered silylated starting material). Spectral data for both compounds were consistent with that reported earlier for 11,15-bis(*O*-*tert*-butyldimethylsilyl) PGE₂ methyl ester.²⁶ However, best results were obtained using the corresponding propargylic iodide which afforded the corresponding readily separable acetylenic PGE₂ derivative **6b** in 51% isolated yield. All yields from this alkylation sequence are unoptimized. The two acetylenic diastereomers could be separated at this stage by column chromatography [2:1 hexane/EtOAc, R_f = 0.16 and 0.18 (PGE₂ stereochemistry)]. Conversion to the corresponding bis(*O*-*tert*-

butyldimethylsilyl) derivative and comparison of ^{13}C NMR spectral data with that reported earlier⁵ confirm the structural assignment.

Preliminary experiments indicate that this highly efficient approach to prostaglandins can be accomplished *in a single step* by sequentially converting the starting diol **3** to a distannyl ether and effecting the organopalladium coupling and alkylation all *in situ* (eq. 6), but the yields at present are less than 25% and further work on this reaction is required.



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References and Footnotes

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- (14) Vinylmercurial **1**: mp 46.5-47.5°C; $[\alpha]_D^{20} = -19.99^\circ$ (c 1.896, HCCl_3); Anal. C, H, Hg; IR (nujol) 1255, 1085, 835, 775 cm^{-1} ; ^1H NMR (DCCl_3) δ 0.02 (s, 3 H, SiCH_3) 0.04 (s, 3 H, SiCH_3), 0.89 (br s, 12 H, t-Bu and CH_3), 1.27 (m, 6 H, CH_2 's), 1.49 (m, 2 H, SiOCHCH_2), 4.14 (dt, 1 H, J = 4.7 and 5.2 Hz, CHOSi), 5.90 (dd, 1 H, J = 17.5 and 4.7 Hz, CH=CHg), 5.98 (d, 1 H, J = 17.5 Hz, = CHHg).
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- (18) This reaction was carried out by adding vinylmercurial 1 (1.05 mmol) to diol 3 (10.5 mmol), Et_3N (3.15 mmol), PdCl_2 (1.05 mmol), and LiCl (2.10 mmol) dissolved in 15 ml of THF at -78°C , allowing the reaction mixture to slowly warm to room temperature for 2 h and stirring at that temperature for 21 h.
- (19) Hydroxyketone 4: R_f 0.31 (2:1 hexane/EtOAc); IR (neat) 3430 (OH), 1748 (C=O), 1245, 1055, 950 cm^{-1} ; ^1H NMR (DCCl_3) δ 0.01 (s, 3 H, SiCH_3), 0.02 (s, 3 H, SiCH_3), 0.89 (br s, 12 H, $t\text{-Bu}$ and CH_3), 1.26 (m, 6 H, CH_2 's), 1.42 (m, 2 H, SiOCHCH_2), 2.08 (dd, 1 H, $J = 18.3$ and 4 Hz, one diastereomer of $\text{O}=\text{CCHCHR}$), 2.11 (dd, 1 H, $J = 18.3$ and 4 Hz, other diastereomer of $\text{O}=\text{CCHCHR}$), 2.23 (dd, 1 H, $J = 18.3$ and 6.9 Hz, $\text{O}=\text{CCHCHOH}$), 2.28 (d, 1 H, $J = 3$ Hz, OH), 2.59 (overlapping dd, 2 H, $J = 18.3$ and 7.5 Hz, CHCOCH), 2.76 (looks like q, 1 H, $J = 6\text{--}7$ Hz, CHOH), 4.07 (looks like q, 1 H, $J = 6$ Hz, CHOSi); ^{13}C NMR (DCCl_3) δ -4.75, -4.29, 13.92, 18.14, 22.50, 24.84, 25.82, 31.67, 38.17, 42.53, 46.11, 47.21, 73.03, 73.81, 128.50, 135.91, 215.32; mass spectrum, m/e 322.2322 [calcd for $\text{C}_{19}\text{H}_{34}\text{O}_2\text{Si}$ (M- H_2O), 322.2328].
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